

## SYNTHETIC METHODOLOGY FOR THE PREPARATION OF *TRANS*- AND *CIS*-2,9-DISUBSTITUTED OXONANES.

Robert W. Carling, Neil R. Curtis and Andrew B. Holmes\*

University Chemical Laboratory, Lensfield Road, CAMBRIDGE CB2 1EW, U.K.

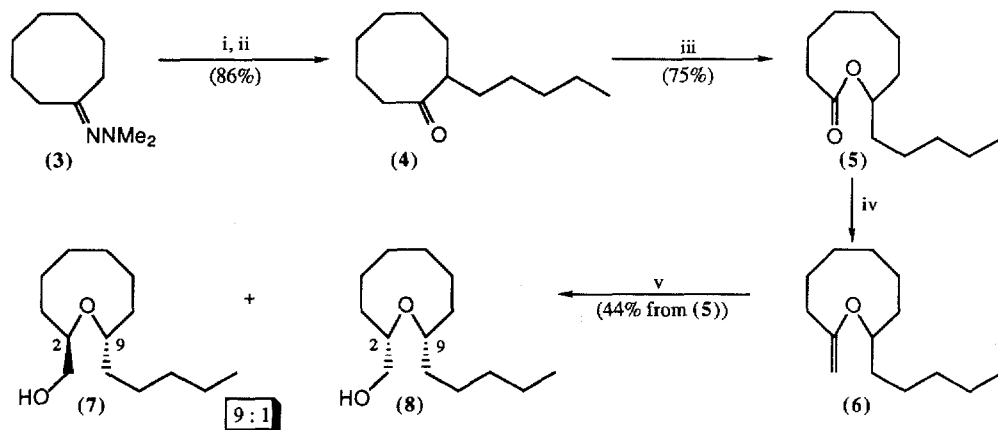
**Summary:** Methylenation of the racemic lactone (5), followed by stereoselective hydroboration, gave predominantly the *trans*-2,9-disubstituted oxonane (7) which was converted into the carbon skeleton (1) of obtusenyne (2). Epimerisation of the *trans*-aldehyde (18) gave the *cis*-compound (19). Relative stereochemistry was established by the asymmetric synthesis of *trans*-(2*R*),(9*R*)-dimethyloxonane (13) and *meso cis*-2,9-dimethyloxonane (17).

Recent isolation of biologically active marine natural products having medium ring ether functionality,<sup>1</sup> predominantly from various *Laurencia* species, has stimulated great interest in developing synthetic methodology for these challenging targets.<sup>2</sup> We have previously reported an approach to the most common oxocane ring size.<sup>3,4</sup> This Letter describes extension of the methodology to the nine-membered, oxonane, ring system having either *trans*- or *cis*-2,9-disubstitution and is illustrated by the synthesis of (1), corresponding to the carbon skeleton of the *Laurencia* natural product obtusenyne (2).<sup>5</sup>



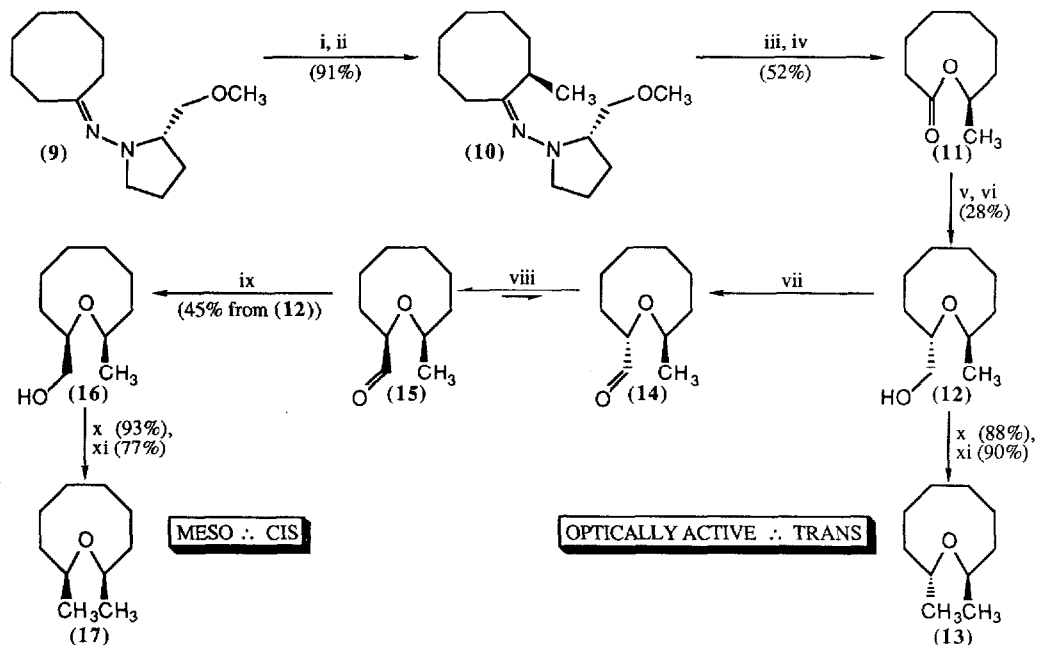
The racemic substituted octanolide (5)<sup>6</sup> was obtained by Baeyer-Villiger oxidation<sup>7</sup> of the ketone (4),<sup>6</sup> available by alkylation of the dimethyl hydrazone (3)<sup>8</sup> (Scheme 1). Methylenation of this lactone with the Tebbe reagent<sup>9</sup> afforded the labile enol ether (6) which was immediately hydroborated with borane-THF complex. Oxidation of the resulting organoborane yielded a 9:1 ratio of *trans*, (7),<sup>6</sup> and *cis*, (8),<sup>6</sup> oxonane alcohols, separable by chromatography. By-products resulting from elimination of the intermediate organoborane were also isolated.

## SCHEME 1



Reagents: i, BuLi, THF,  $-78^{\circ}\text{C}$  then  $\text{C}_5\text{H}_{11}\text{I}$ ; ii, 10% aq.  $\text{H}_2\text{SO}_4$ ; iii,  $\text{CF}_3\text{CO}_3\text{H}$ ,  $\text{Na}_2\text{HPO}_4$ ,  $\text{CH}_2\text{Cl}_2$ ; iv,  $(\text{C}_5\text{H}_5)_2\text{Ti}(\mu\text{-CH}_2, \mu\text{-Cl})\text{AlMe}_2$ , DMAP, THF, toluene,  $-40^{\circ}$  to  $25^{\circ}\text{C}$  then NaOH,  $-15^{\circ}$  to  $25^{\circ}\text{C}$ ; v,  $\text{BH}_3$ -THF, THF,  $0^{\circ}\text{C}$  then NaOH,  $\text{H}_2\text{O}_2$ ,  $0^{\circ}\text{C}$ .

## SCHEME 2

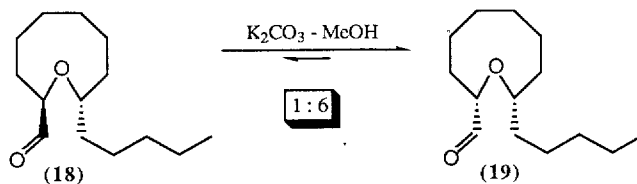


Reagents: i, LDA, THF,  $0^{\circ}\text{C}$ , 15h; ii, MeI,  $-95^{\circ}\text{C}$ ; iii,  $\text{O}_3$ ,  $\text{CH}_2\text{Cl}_2$ ,  $-78^{\circ}\text{C}$ ; iv,  $\text{CF}_3\text{CO}_3\text{H}$ ,  $\text{Na}_2\text{HPO}_4$ ,  $\text{CH}_2\text{Cl}_2$ ; v,  $(\text{C}_5\text{H}_5)_2\text{Ti}(\mu\text{-CH}_2, \mu\text{-Cl})\text{AlMe}_2$ , DMAP, THF, toluene,  $-40^{\circ}$  to  $25^{\circ}\text{C}$  then NaOH,  $-15^{\circ}$  to  $25^{\circ}\text{C}$ ; vi,  $\text{BH}_3$ -DMS, THF,  $0^{\circ}\text{C}$  then NaOH,  $\text{H}_2\text{O}_2$ ,  $0^{\circ}\text{C}$ ; vii, PCC,  $\text{CH}_2\text{Cl}_2$ ; viii,  $\text{K}_2\text{CO}_3$ , MeOH, 3d; ix,  $\text{NaBH}_4$ ; x, TsCl, DMAP,  $\text{CH}_2\text{Cl}_2$ ; xi,  $\text{LiAlH}_4$ , ether.

The presence of an nOe between H<sub>2</sub> and H<sub>9</sub> in the 3,5-dinitrobenzoate of (8), and no such effect in the corresponding derivative of (7), supported the assigned relative stereochemistry. However, limited information about nine-membered ring conformations precluded unequivocal assignment. Furthermore, the apparent contrast between *trans*-selective hydroboration of (6) and the corresponding preference for *cis*-compounds in the oxocane<sup>3,4,10</sup> and oxepane<sup>11</sup> series demanded proof of relative stereochemistry.

In order to establish unambiguously the validity of the assigned relative stereochemistry, asymmetric synthesis of *trans*-(2*R*),(9*R*)-dimethyloxonane (13) and *meso cis*-2,9-dimethyloxonane (17) was undertaken (Scheme 2).<sup>13</sup>

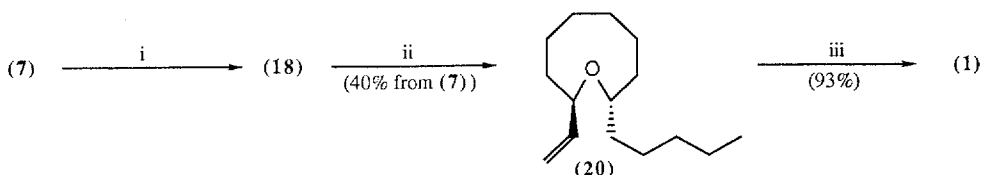
Ozonolysis followed by Baeyer-Villiger oxidation<sup>14</sup> of SAMP-hydrazone (10)<sup>15</sup> gave octanolide (11)<sup>6</sup> {[α]<sub>D</sub> -41° (*c* 1.0, MeOH)} in 91% e.e.<sup>16</sup> Methylenation followed by hydroboration-oxidation gave a single oxonane product (12)<sup>6</sup> {[α]<sub>D</sub> -23° (*c* 1.1, CHCl<sub>3</sub>)}. This was converted into the corresponding tosylate then reduced to give *trans*-(2*R*),(9*R*)-dimethyloxonane (13)<sup>6</sup> {[α]<sub>D</sub> -50° (*c* 0.51, CDCl<sub>3</sub>)}. The *cis*-oxonane alcohol (16)<sup>6</sup> {[α]<sub>D</sub> -16° (*c* 0.6, CHCl<sub>3</sub>)} was available from (12) by oxidation to the aldehyde, base-catalysed epimerisation, and reduction [49:1 ratio of (16):(12) by G.C.]. Conversion by the same sequence as before gave *cis*-2,9-dimethyloxonane (17)<sup>6</sup> {[α]<sub>D</sub> 0° (*c* 0.59, CDCl<sub>3</sub>)}.<sup>17</sup> This confirmed that the major product from the methylenation-hydroboration sequence on a 9-substituted octanolide did indeed have *trans* relative stereochemistry.



Base-catalysed equilibration of the racemic aldehyde (18) favoured (6:1) the *cis*-product (19). Thus, a versatile strategy for synthesis of both *trans* and *cis*-2,9-disubstituted oxonanes has been developed.

Synthesis of the carbon skeleton (1) of obtusenyne (2) from the alcohol (7) was achieved as outlined in Scheme 3. Oxidation followed by Wittig homologation gave the alkene (20),<sup>6</sup> which was hydrogenated to give *trans*-2-ethyl-9-pentyloxonane (1).<sup>6</sup>

SCHEME 3

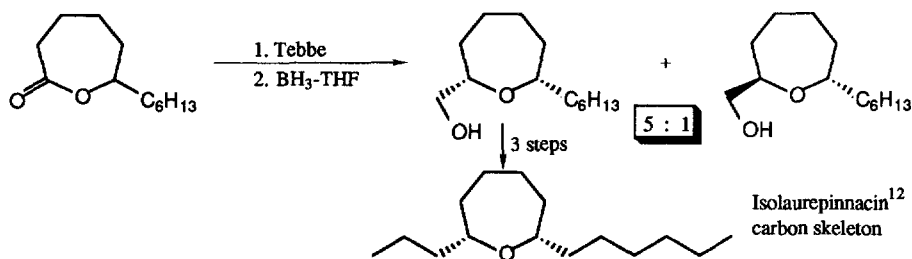


Reagents : i, PCC, CH<sub>2</sub>Cl<sub>2</sub>; ii, Ph<sub>3</sub>P=CH<sub>2</sub>, THF; iii, H<sub>2</sub>, Pd-C, EtOAc.

We thank the S.E.R.C. for studentships (to R.W.C. and N.R.C.), Merck, Sharp & Dohme (Harlow) for financial support, Adrian Smith for the chiral shift study, and Peter Hamley for molecular modelling.

## REFERENCES AND FOOTNOTES

1. R.E. Moore in *Marine Natural Products*, P.J. Scheuer, Ed., Academic Press, 1978, Vol. 1, Ch. 2, p. 43; K.L. Erickson, *ibid.*, 1983, Vol. 5, Ch. 4, p. 131; D.J. Faulkner, *Nat. Prod. Rep.*, 1984, **1**, 251; 1986, **3**, 1; 1987, **4**, 539; 1988, **5**, 613.
2. L.E. Overman and A.S. Thompson, *J. Am. Chem. Soc.*, 1988, **110**, 2248; K.C. Nicolaou, C.V.C. Prasad, C.-K. Hwang, M.E. Duggan, and C.A. Veale, *ibid.*, 1989, **111**, 5321, and pertinent references cited in these recent examples.
3. R.W. Carling and A.B. Holmes, *J. Chem. Soc., Chem. Commun.*, 1986, 565; *Tetrahedron Lett.*, 1986, **27**, 6133.
4. J.S. Clark and A.B. Holmes, *Tetrahedron Lett.*, 1988, **29**, 4333.
5. T.J. King, S. Imre, A. Öztunc, and R.H. Thomson, *Tetrahedron Lett.*, 1979, 1453; B.M. Howard, G.R. Schulte, W. Fenical, B. Solheim, and J. Clardy, *Tetrahedron*, 1980, **36**, 1747.
6. All new compounds exhibited satisfactory spectroscopic, combustion analysis and/or high resolution mass spectral data.
7. W.C. Still and I. Galynker, *Tetrahedron*, 1981, **37**, 3981.
8. H. Hart, B.-L. Chen, and M. Jeffares, *J. Org. Chem.*, 1979, **44**, 2722.
9. F.N. Tebbe, G.W. Parshall, and G.S. Reddy, *J. Am. Chem. Soc.*, 1978, **100**, 3611; S.H. Pine, R. Zahler, D.A. Evans, and R.H. Grubbs, *ibid.*, 1980, **102**, 3270; K.A. Brown-Wensley, S.L. Buchwald, L. Cannizzo, L. Clawson, S. Ho, D. Meinhardt, J.R. Stille, D. Straus, and R.H. Grubbs, *Pure Appl. Chem.*, 1983, **55**, 1733, and references cited therein.
10. The *cis*-selective hydroboration of 2-methyleneoxocanes is consistent with the approach of reagent to the calculated<sup>7</sup> minimum energy conformation.<sup>4</sup> The corresponding prediction for the oxonanes is ambiguous, and requires further study.
11. R.W. Carling, *Ph. D. Thesis*, University of Cambridge, 1986.



12. A. Fukuzawa and T. Masamune, *Tetrahedron Lett.*, 1981, **22**, 4081.
13. A similar approach was used to establish the stereochemistry of an oxocene product: L.E. Overman, T.A. Blumenkopf, A. Castañeda, and A.S. Thompson, *J. Am. Chem. Soc.*, 1986, **108**, 3516.
14. This one-pot procedure<sup>4</sup> avoided racemisation of the ketone intermediate and exposure to carcinogenic nitrosamines.
15. D. Enders and H. Eichenauer, *Chem. Ber.*, 1979, **112**, 2933.
16. The enantiomeric excess (e.e.) was determined on the methanolysis product methyl (8*R*)-hydroxynonanoate by <sup>1</sup>H NMR chiral shift study using (+)-Eu(hfc)<sub>3</sub>.
17. In addition, [α]<sup>0</sup> at λ 578, 546, 436, and 365nm.

(Received in UK 18 September 1989)